2020 - 2021

DECAY LAW

No is the number of radioactive atoms/Nuclei at t=0 N is the number of radioactive atoms/Nuclei left after time t Larger the value of N larger the decay Let dN be the number of decays

in time dt

 $dN \alpha - N dt$ $dN = -\lambda N dt$

where λ is constant of proportionality called the decay constant.

The negative sign shows that the change in nuclei dN is negative.

No

N₀ 2

 $\frac{N_0}{4}$

 $N = N_0 e^{-\lambda t}$

 $T_{\frac{1}{2}}$

2T1

$$\frac{dN}{N} = -\lambda dt$$
Integrating we get, $\int_{N_0}^{N} \frac{dN}{N} = -\lambda \int_{0}^{t} dt$
 $[\ln N]_{N_0}^{N} = -\lambda [t]_{0}^{t}$

$$\ln N - \ln No = -\lambda [t - 0]$$
$$\ln \frac{N}{No} = -\lambda t$$

 $N = Noe^{-\lambda t}$ This is the decay law or law of radioactive decay.

ACTIVITY

Activity A, is rate of disintegration or disintegrations per unit time $A = -\frac{dN}{dt} = -\lambda N = -\lambda No e^{-\lambda t}$

 $A = Ao e^{-\lambda t}$ where $Ao = -\lambda No$

Activity is measure in becquerel (Bq) in SI unit One becquerel is one disintegration per second 1 Curie (Ci) = 3.7 x 10¹⁰ Bq

HALF LIFE

The time taken for the number of parent radioactive nuclei of a oarticular species to reduce to half its value is called its half – life $(T_{\frac{1}{2}})$

 $N = No e^{-\lambda t}$ No Put N as No $\frac{No}{2}$ = No e $\lambda T_{\frac{1}{2}}$ $2 = e^{\prime}$ $= N_0 e^{-\lambda t}$ $\ln 2 = \lambda T_1$ N₀ 0.693 $T_{1/2} = -$ No So in time $t=T_{\frac{1}{2}}$ the radioactive parent nuclei reduce to No/2 0 2T1 $T_{\frac{1}{2}}$ In time t= $2T_{\frac{1}{2}}$ the radioactive parent nuclei reduce to No/4 and so on

AVERAGE LIFE

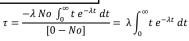
Average or mean life = Total lifetime of all nuclei Total number of nuclei in sample

$$\tau = \frac{\int_{N_0}^0 t \, dN}{\int_{N_0}^0 dN} = \frac{-\lambda \, No \, \int_0^\infty t \, e^{-\lambda t} \, dt}{[N]_{N_0}^0}$$

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On integrating we get

$$\tau = \frac{1}{\lambda}$$

Thus, decay constant is also the reciprocal of mean or average life of the radioactive species.

